

10-Benzyl-3,3,6,6,9-pentamethyl-3,4,6,7,9,10-hexahydroacridine-1,8(2H,5H)-dione

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Key indicators

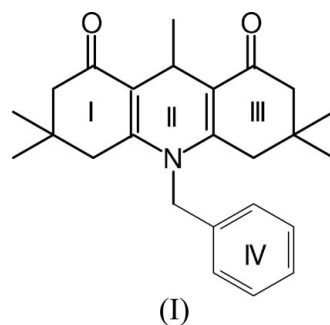
Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$
Disorder in main residue
 R factor = 0.065
 wR factor = 0.197
Data-to-parameter ratio = 13.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{C}_{25}\text{H}_{31}\text{NO}_2$, crystallizes with two molecules in the asymmetric unit. In both molecules, the central dihydropyridine ring adopts a flattened boat conformation. In molecule *A*, the outer rings adopt sofa conformations, whereas in molecule *B*, one of the outer rings is disordered and the other ring adopts a sofa conformation. In the crystal structure, $\text{C}-\text{H}\cdots\text{O}$ interactions link the molecules to form a tape that runs along the *b* axis.

Comment

Acridine and its derivatives exhibit a wide spectrum of biological activities, such as mutagenic, antitumour (Talacki *et al.*, 1974), anti-amoebic (Prasad Krishna *et al.*, 1984), hypertensive and anti-inflammatory (Asthana *et al.*, 1991). Acridine-containing drugs have been found to possess antiprotozoal activity (Karolak-Wojciechowska *et al.*, 1996) and are used for the treatment of Alzheimer's disease (Bandoli *et al.*, 1994). Substituted hexahydroacridine-1,8-diones resemble K-channel openers, which relax KCl-preconcentrated urinary-bladder muscle *in vitro* (Trivedi *et al.*, 1995; Li *et al.*, 1996). Acridine-1,8-diones exhibit fluorescence and laser activities (Selladurai *et al.*, 1990). Decahydroacridine-1,8-diones act as photosensitizers (Timpe *et al.*, 1993) and also possess other important photophysical and electrochemical properties (Mohan *et al.*, 1996). Acridinediones were found to have laser activity around 475–495 nm (Murugan *et al.*, 1998). The present study of the title compound, (I), is part of a series of investigations on the crystal structures of acridinedione derivatives.



There are two molecules in the asymmetric unit. As reported in related acridinedione derivatives (Ganesh, Velmurugan *et al.*, 1998; Ganesh, Banumathi *et al.*, 1998; Ganesh *et al.*, 1999; Sankaranarayanan *et al.*, 1998, 1999; Jeyakanthan *et al.*, 2000, 2002; Aravindan *et al.*, 2003), the acridine system is folded about the line passing through atoms C9 and N10, with a dihedral angle of 18.9 (1) and 18.1 (1)° for

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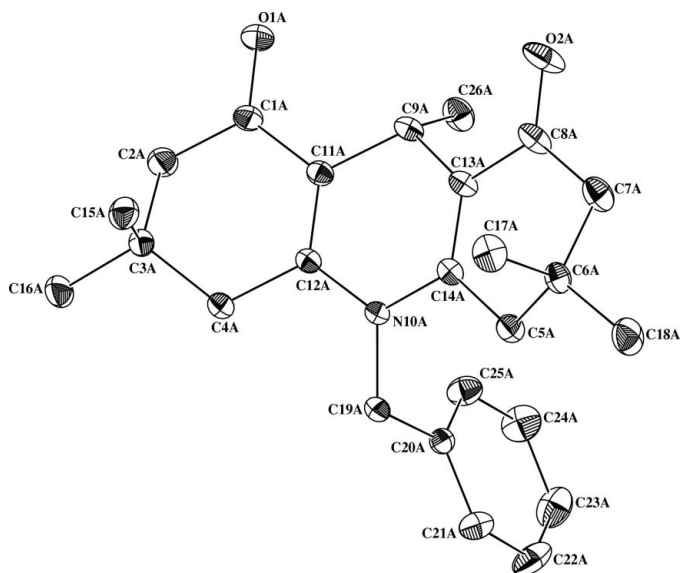


Figure 1
One of the two independent molecules of the title compound, showing 35% probability displacement ellipsoids. H atoms have been omitted.

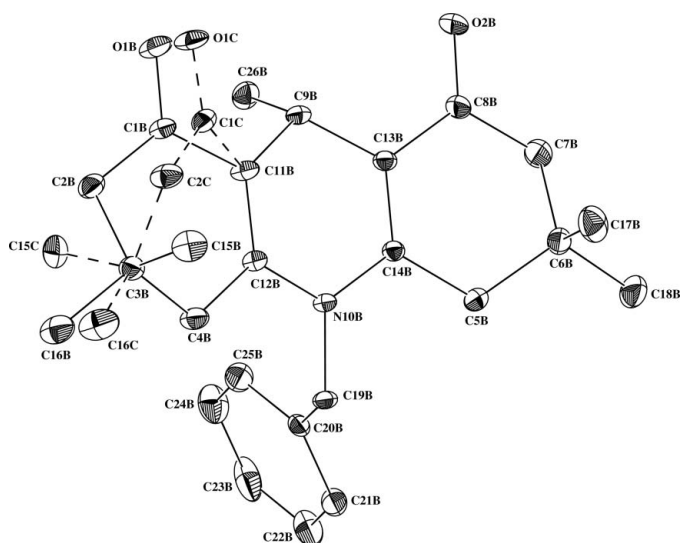


Figure 2
The other independent molecule of the title compound, showing 35% probability displacement ellipsoids. H atoms have been omitted. Both disorder components are shown.

molecules *A* and *B*, respectively, between the two halves of the molecule (C9/C11/C1/C2/C4/C12/N10 and C9/C13/C8/C7/C5/C14/N10 for molecule *A*, and C9/C11/C3/C4/C12/N10 and C9/C13/C8/C7/C5/C14/N10 for molecule *B*). The dihedral angles between the least-squares plane through the central dihydropyridine (C11/C12/C13/C14) and phenyl rings (C20–C25) are 81.8 (1) and 84.2 (1)° for molecule *A* and *B*, respectively. The methyl groups attached to the dihydropyridine rings are in a pseudo-axial position.

The puckering of the dihydropyridine ring is quite small, owing to the π conjugation in the C11–C12–N10–C14–C13 system. Similar features have also been observed in other related acridinedione derivatives (Gunasekaran *et al.*, 1996; Ganesh, Banumathi *et al.*, 1998; Ganesh *et al.*, 1999; Sankar-

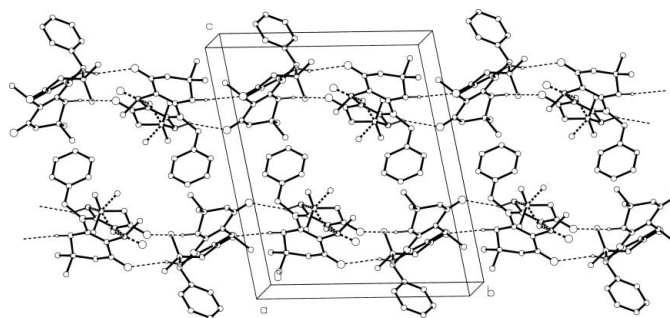


Figure 3
The molecular packing of (I), viewed approximately down the *a* axis, showing the C–H···O-mediated tape motif along the *b* axis. Hydrogen bonds are shown as dashed lines.

anarayanan *et al.*, 1998; Aravindan *et al.*, 2003). The sum of the bond angles around N10 (358.7 and 359.9° for molecules *A* and *B*, respectively) indicates sp^2 -hybridization.

In the crystal structure, C–H···O interactions (Table 2) link molecules *A* and *B* to form tapes that run along *b* axis (Fig. 3).

Experimental

A mixture of 2,2'-ethylidenebis(5,5-dimethylcyclohexane-1,3-dione) (1.65 g, 5.3 mmol) and benzylamine (0.5 g, 5.3 mmol) was refluxed in acetic acid (20 ml) for 6 h. The reaction mixture was cooled and poured on to crushed ice. The resulting solid was filtered off and purified by column chromatography over alumina (neutral) and eluted with CHCl_3 –MeOH (2:1), to give the title compound. Single crystals were grown by slow evaporation of a solution in CHCl_3 –MeOH (1:1).

Crystal data

$\text{C}_{25}\text{H}_{31}\text{NO}_2$
 $M_r = 377.51$
Triclinic, $P\bar{1}$
 $a = 10.3359$ (1) Å
 $b = 13.8339$ (3) Å
 $c = 16.2254$ (4) Å
 $\alpha = 101.153$ (1)°
 $\beta = 90.934$ (1)°
 $\gamma = 108.687$ (1)°
 $V = 2148.53$ (8) Å³

$Z = 4$
 $D_x = 1.167$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 4761 reflections
 $\theta = 1.3$ – 28.4 °
 $\mu = 0.07$ mm⁻¹
 $T = 293$ (2) K
Plate, yellow
 $0.46 \times 0.24 \times 0.08$ mm

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.967$, $T_{\max} = 0.994$
11586 measured reflections

7331 independent reflections
4386 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$
 $\theta_{\text{max}} = 25.0$ °
 $h = -12 \rightarrow 10$
 $k = -16 \rightarrow 16$
 $l = -19 \rightarrow 16$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.065$
 $wR(F^2) = 0.197$
 $S = 1.03$
7331 reflections
563 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0876P)^2 + 0.573P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.39$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

O1A—C1A	1.229 (3)	O1C—C1C	1.223 (19)
O2A—C8A	1.232 (4)	O2B—C8B	1.241 (4)
N10A—C14A	1.399 (3)	N10B—C12B	1.399 (3)
N10A—C12A	1.404 (3)	N10B—C14B	1.402 (4)
C11A—C12A	1.349 (4)	C11B—C12B	1.359 (4)
C13A—C14A	1.350 (4)	C13B—C14B	1.359 (4)
O1B—C1B	1.225 (13)		
C14A—N10A—C12A	119.4 (2)	C12B—N10B—C14B	119.8 (2)
C14A—N10A—C19A	119.6 (2)	C12B—N10B—C19B	121.3 (2)
C12A—N10A—C19A	120.7 (2)	C14B—N10B—C19B	118.8 (2)

Table 2

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
C4A—H4A1...O2B	0.97	2.47	3.369 (4)	153
C5A—H5A2...O1B	0.97	2.35	3.291 (12)	163
C4B—H4B1...O2A ⁱ	0.97	2.48	3.362 (4)	152
C5B—H5B2...O1A ⁱ	0.97	2.39	3.324 (4)	163

Symmetry code: (i) $x, y - 1, z$.

All H atoms were positioned geometrically and allowed to ride on their parent atoms with C—H = 0.93–0.98 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{C})$. In molecule *B*, atoms C1B and C2B, as well as the carbonyl atom O1B and methyl groups C15B and C16B, were found to be disordered, and the site-occupancy factors of the major and minor conformations were refined to 0.591 (8):0.409 (8)].

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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